Analysis of seized stanozolol formulations in South Brazil by liquid chromatography coupled to quadrupole time-of-flight-mass spectrometry

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Anabolic-androgenic steroids (AAS) are synthetic derivatives of testosterone which are used medically for several diseases. However, misuse is commonly observed by athletes to promote enhancement of strength, bodyweight and performance. AAS are frequently obtained by internet black market from clandestine drug manufacturing laboratories, without any quality standards, being potentially dangerous for users. The purpose of this work was the development and application of a fast and simple procedure for the quantitation of stanozolol by liquid chromatography coupled to quadrupole time-of-flight-mass spectrometry (LC-QTOF-MS) in tablets packs seized in Rio Grande do Sul state, Brazil. The samples of stanozolol were separated considering its pharmaceutical forms. An internal standard was added to the aliquots of the samples, dissolved in methanol and 5 µL were injected into the analytical system. The newly developed method has been validated for lower limit of quantitation (LLOQ), linearity, accuracy, precision and selectivity. The lower limit of quantification (LLOQ) was 0.1 µg/mL. The developed method was successfully applied to 31 samples seized by the *Secretaria da Receita Federal do Brasil* (a Brazilian federal revenue service agency). According to the results, 90.3% of the suspected medicines (n=31) were adulterated, and 65% exhibited higher concentrations of stanozolol than those indicated in the label. This work successfully established a new method for quantification of stanozolol using LC-QTOF-MS. This method aims at contributing to the identification and quantification of this anabolic androgenic steroid frequently seized by federal inspection agencies.

Keywords: Anabolic-androgenic steroids, stanozolol, liquid chromatography, mass spectrometry.

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Introduction

Anabolic-androgenic steroids (AAS) are synthetic derivatives of testosterone and they are used medically for the treatment of hypogonadism, metabolic disorders, burns, anemia, bone marrow failure, chronic renal failure and acquired immunodeficiency syndrome (1,2). However, misuse is commonly observed by athletes to promote strength increase, bodyweight and performance gain (1,3). Adverse events were associated with a variety of different cardiovascular diseases (4-6), liver disorders including peliosis hepatis, hepatocellular hyperplasia and carcinomas (7) and a spectrum of behavioral effects correlated with hyperandrogenism (8).

Legitimate pharmaceutical grade AAS are illegally acquired in pharmacies. In addition, black marketed AAS have frequently been obtained by internet from clandestine drug manufacturing laboratories, without any quality standards. In both cases, these products can be potentially harmful for users promoting intoxication (9,10).

In Brazil, studies indicate that erectile dysfunction medicines, AAS and anorexigenic products are the most commonly counterfeit and contraband drugs (11). Brazilian Federal Police reported that, from 2006 to 2011, one third of the 3,537 AAS analyzed products were

counterfeit products (12). Additionally, Neves et al. conducted the analysis of 345 AAS positive medicines/dietary supplements samples and observed that 53% were substandard, spurious, falsely labelled, falsified or counterfeit (SSFFC) medical products (13). The measurements conducted by Ribeiro et al. indicated that 20% of seized AAS samples analyzed showed adulterants not described in the labels. Among those, stanozolol samples presented the most unusual results since none of the capsules had the analyte presence, showing adulteration with other AAS compounds and sildenafil citrate (14).

Stanozolol (STZ) is an alkylated C17 α steroid commonly detected SSFFC medical product in Brazilian scenario (9). It is a heterocyclic steroid derived from dihydrotestosterone, containing chemical alterations in order to increase anabolic activity and decrease androgenic pathways, resulting in and anabolic/androgenic ratio of 320/30 (15,16). Because of its anabolic characteristics, STZ can stimulate protein and erythropoietin synthesis. Therefore, it promotes body fat loss while preserving the lean mass to achieve the physical shape desired (17). Despite all the beneficial outcomes, STZ presents side effects such as most of the AAS, even with their low androgenicity. Voice changes, alopecia, hair growth, clitoris increase, and many others

have been reported. STZ particularly exhibits a high hepatotoxic activity, since the presence of the alkyl in C17- α reduces liver metabolism.

Stanozolol analytical measurements can be conducted by several techniques. This compound can be determined in several biological matrices such as blood, plasma, serum, urine, hair, nails and oral fluid for doping control purposes (18). Also, it can be evaluated in pharmaceutical formulations in order to determine its authenticity. In this scenario, chromatographic approaches are the most established methodologies (14). Gas chromatography coupled to mass spectrometry (MS) is widely used for screening, while liquid chromatography coupled to MS or tandem mass spectrometry (MS/MS) ensures the required analysis precision, specificity and sensibility with minimal sample pretreatment (18,19). Moreover, spectroscopic methods such as proton nuclear magnetic resonance (^1H NMR), Raman and Fourier transform infrared spectroscopy (FT-IR) can be applied to the identification and quantitation of stanozolol (14,20). In addition, the use of high-resolution mass spectrometry (HRMS) methods is suggested due to the ability to improve mass accuracy and enhance the selectivity of the target compounds and/or allow for the analysis of untargeted compounds.

The aim of the present study was to evaluate the quality of formulations containing the anabolic steroid stanozolol, seized by the *Secretaria da Receita Federal do Brasil* (a Brazilian federal revenue service agency). For this purpose, a liquid chromatography coupled to quadrupole time-of-flight-mass spectrometry (LC-QTOF-MS) method was developed and validated, being applied to the analysis of 31 stanozolol formulations.

Materials and Methods

Chemical and reagents

Methanolic solutions of stanozolol and methyltestosterone (internal standard; IS) at a concentration of 1.0 mg/mL were obtained from Cerilliant Analytical Reference Standards (Round Rock, TX, USA). HPLC-grade methanol, acetonitrile and formic acid were purchased from Merck (Darmstadt, Germany). Water was purified using a Milli-Q system (Millipore, Billerica, MA, USA). All reference standards were > 99% purity.

Preparation of standard solutions

Stock solutions of stanozolol and IS were prepared at concentrations of 200 $\mu\text{g/mL}$ and 10 $\mu\text{g/mL}$ respectively, using methanol. These solutions were stored under refrigeration (2-8 $^{\circ}\text{C}$). Working solutions with concentrations between 0.1 and 20 $\mu\text{g/mL}$ were also prepared in methanol by diluting stock solutions.

Chromatographic conditions and data acquisition

A micrOTOF-Q III (Bruker Daltonics, Bremen, Germany) coupled to a Shimadzu Prominence liquid

chromatography system (Shimadzu, Kyoto, Japan) was used for the analyses. Mass data were obtained with the following optimized conditions: capillary voltage, 3000 V; nebulizer gas, 3 bar; drying gas, 5 L/min; drying temperature, 150 $^{\circ}\text{C}$; quadrupole ion energy, 4 eV; mass range, m/z 50 - 350 acquired with 1 Hz. The system was calibrated in positive mode using a solution of sodium formate (10 mM) and the protonated molecular ions m/z 329.2584 (stanozolol) and m/z 303.2325 (methyltestosterone) were detected with a mass error of < 5 ppm. The following chromatographic condition was used: a 75 $\times$ 2.0 mm i.d., 2.2 μm , Shim-pack XR-ODS II (Shimadzu, Kyoto, Japan) column eluted with a gradient of 0.1% formic acid in water (A) and acetonitrile (B) at a 400 $\mu\text{L/min}$ flow rate and 45 $^{\circ}\text{C}$ as follows: 0 – 2.5 min, 40 – 100% of B; 2.5 – 3.5 min, 100% of B; 3.5 – 3.6 min, 100 – 40% of B; 3.6 – 5 min, 40% of B.

Sample preparation and analyses

Analyses were performed using stanozolol formulations samples seized by the *Secretaria da Receita Federal do Brasil*. Sample classification was executed in order to discriminate tablet from capsules, resulting in 31 packs ($n=29 \times 100$ tablets; $n=2 \times 50$ capsules). The tablets packs were manufactured by a Paraguayan laboratory which has no health license in Brazil. The capsule packs exhibited no manufacturing nor expiration dates, only a handwritten identification of the active ingredient, concentration and number of capsules. The means of the weights of each batch were determined through the individual measurements of 20 tablets and capsules (with and without the content). For the individual weight evaluation, one tablet/capsule of each pack was weighted and compared to the means obtained. This procedure was carried out according to the recommendations of the 32nd edition of the American Pharmacopeia (USP 43) (21). Afterwards, a 1 mg were weighed of the tablets/capsules on an analytical balance (Shimadzu, AUW220D; 0.1mg/0.01mg) was transferred to falcon tubes and diluted with 4 mL of methanol, followed by homogenization in vortex for 30 s and centrifugation for 10 min at 3500 rpm. For quantitative analysis IS working solution (100 $\mu\text{g/mL}$) was added to all samples and an aliquot of 5 μL was directly injected into the analytical system.

Optimization of the Q-TOF parameters

In this step, a multivariate approach strategy based on a central composite design was applied for the optimization of three parameters that are relevant in order to obtain the maximum analytical response of the target compound: capillary voltage (1320 V; 2200 V; 3500 V; 4800 V; 5680 V), drying temperature (150 $^{\circ}\text{C}$; 190 $^{\circ}\text{C}$; 250 $^{\circ}\text{C}$; 310 $^{\circ}\text{C}$; 350 $^{\circ}\text{C}$) and collision energy (1 eV; 6 eV; 13 eV; 20 eV; 25 eV). A total of 17 experiments were evaluated using as response the absolute area provided by the analyte in all examined conditions. The software *Statistica* (Statsoft,

EUA) was used to evaluate the experimental design through a response surface methodology.

Validation

The method was validated according to international guidelines and recommendations for validation of analytical methods, by establishing parameters such as the lower limit of quantitation (LLOQ), linearity, accuracy, precision and selectivity (22-25). The study of linearity was performed by injecting working standard solutions of the analytes at six levels: 0.1, 1, 5, 10, 15 and 20 µg/mL, in six replicates for each concentration. Precision and accuracy were performed through the analysis of three different levels (0.2, 8 and 16 µg/mL). Six replicates on each of three consecutive days were acquired and the precision data (within and between-day) were calculated using one-way analysis of variance (ANOVA). Selectivity was evaluated in the lowest level of precision, adding compounds that may be present in the sample (ephedrine, caffeine and sibutramine).

Results and Discussion

Optimization of the Q-TOF parameters

The response surfaces obtained with the variable's capillary voltage, temperature and collision energy are shown in Figure 1. According to the results, the optimized results of capillary voltage, temperature and collision energy are respectively 3000 V, 150 °C and 4 eV. These values represent the highest analytical signal intensity for stanozolol, and they were used for subsequent validation and analyses. The use of multivariate approaches to determine the optimal conditions in experimental procedures is a valid and effective strategy. Traditional univariate approaches can be comprised of several empirical experiments, which can make the analysis time-consuming and more expensive.

Figure 1. Response surfaces (peak area) for the evaluation of the Q-TOF parameters: a) Capillary Voltage (V) x Temperature (°C); b) Collision Energy (eV) x Capillary Voltage (V); c) Collision Energy (eV) x Temperature (°C).

Validation

The LLOQ value was determined by empirical method, which consists of analyzing a series of samples concentrations of the analyte, and the LLOQ obtained was 0.1 µg/mL (Figure 2). The calibration curve was linear over the specified range (0.1 – 20 µg/mL). The linear regression equation and the respective coefficient of determination were $y = 0.0018x + 0.1653$, $r^2 = 0.9975$. The analyte has presented heteroscedastic calibration curve,

which may be explained by the wide range of concentration values considered in this parameter. Therefore, linear least squares regression methods have resulted in large errors in the calculation of the concentrations, especially at lower values. By applying weighted least squares linear regression, the resulting weighting factor was of 1/y and the linear regression equation applied was $y = 1.141319x - 0.09435$ ($r^2 = 0.9919$).

Figure 2. Chromatogram of stanozolol and its respective internal standard (methyltestosterone) at LLOQ value (0.1 µg/mL).

The method selectivity was performed under the specified test conditions and no significant level of interfering at the retention time of the analyte was observed, thus confirming the selectivity of the method. The precision was assessed by ANOVA and the results obtained of intra-day precision presented a variation of 11.4 - 17.8%. For inter-day precision the results were achieved in the range of 11.8 to 16.7%. Results on precision and accuracy of the validated method are shown in Table 1.

Table 1. Precision and accuracy data of the developed method for stanozolol formulations at three quality control (QC) levels: QC Low (0.2 µg/mL), QC Medium (8 µg/mL), QC High (16 µg/mL).

Parameters	Quality Controls (QC)	Results
Accuracy (%)	QC Low	92.67
	QC Medium	95.24
	QC High	92.10
Intra-day precision (% CV)	QC Low	17.79
	QC Medium	12.05
	QC High	11.04
Inter-day precision (% CV)	QC Low	16.68
	QC Medium	13.88
	QC High	13.69

CV: Coefficient of variation.

Samples analysis

The means of the weight of tablets and capsules, so as the individual weights of each sample and the variation are summarized in Table 2. It was noted that 5 samples (4 tablets and 1 capsule) exceeded the variation limit of $\pm 5\%$ for tablets and $\pm 7.5\%$ for capsules established by the regulatory agencies. These variations can directly influence on the content of stanozolol in the product compromising its quality.

Table 2. Quality control data for mean weight calculated by batch, individual weights and variation. NM: Not mentioned

Batch	Sample number	Batch weight (mg)	Mean weight (mg)	Weight variation (mg)	Weight variation (%)
Tablets					
16116	14	252.50	249.55	2.95	1.18%
16116	15	266.81		17.26	6.92%
16116	18	240.09		-9.46	-3.79%
16116	19	257.92		8.37	3.35%
16116	21	232.87		-16.68	-6.68%
16116	23	263.59		14.04	5.63%
16116	24	255.95		6.40	2.56%
16116	26	250.59	250.25	1.04	0.42%
16116	27	257.62		8.07	3.23%
16098	1	257.32	250.25	7.07	2.83%
16098	22	244.54		-5.71	-2.28%
16097	4	250.48	255.63	-5.15	-2.01%
16074	12	259.04	249.92	9.12	3.65%
16074	13	245.62		-4.30	-1.72%
16063	2	242.85	254.45	-11.60	-4.56%
16063	7	234.15		-20.30	-7.98%
16063	9	261.60		7.15	2.81%
16063	10	265.55		11.10	4.36%
16046	25	258.84	258.23	0.61	0.24%
16045	3	258.01	255.58	2.43	0.95%
16045	11	257.00		1.42	0.56%
16045	16	257.78		2.20	0.86%
16045	28	255.12		-0.46	-0.18%
16043	6	266.10	256.87	9.23	3.59%
16041	5	244.31	250.94	-6.63	-2.64%
16041	20	254.53		3.59	1.43%
16017	8	248.65	250.01	-1.36	-0.54%
16013	17	258.70	256.26	2.44	0.95%
16013	29	258.41		2.15	0.84%
Capsules					
NM	30	378.78	393.53	-14.75	-3.75%
NM	31	378.87	413.18	-34.31	-8.30%

After the validation of the LC-QTOF-MS method, the concentration of the active ingredient of stanozolol in each tablet and capsule was calculated. The reference parameter of variation of the active ingredient concentration was the monograph of stanozolol described in the 32nd edition of the American Pharmacopeia (21). This document defines that stanozolol tablets should not contain less than 90.0% or more than 110.0% of the labeled amount. Quantitative results for each sample analyzed are shown in Table 3. Variation of the content measurements are presented in the control graph (Figure 3).

Table 3. Concentration results for the analyses of stanozolol formulations.

Batch	Sample number	Concentration indicated on the label (mg)	Individual weight (mg)	Concentration measurements by LC-HRMS (mg)
Tablets				
16116	14	10	252.50	11.72
16116	15	10	266.81	13.85
16116	18	10	240.09	10.56
16116	19	10	257.92	16.95
16116	21	10	232.87	14.79
16116	23	10	263.59	11.64
16116	24	10	255.95	12.38
16116	26	10	250.59	11.71
16116	27	10	257.62	11.01
16098	1	10	257.32	10.65
16098	22	10	244.54	13.04
16097	4	10	250.48	8.06
16074	12	10	259.04	8.14
16074	13	10	245.62	14.34
16063	2	10	242.85	22.49
16063	7	10	234.15	7.70
16063	9	10	261.60	8.12
16063	10	10	265.55	13.99
16046	25	10	258.84	13.28
16045	3	10	258.01	12.98
16045	11	10	257.00	13.56
16045	16	10	257.78	7.54
16045	28	10	255.12	13.57
16043	6	10	266.10	4.27
16041	5	10	244.31	10.33
16041	20	10	254.53	18.16
16017	8	10	248.65	8.33
16013	17	10	258.70	13.48
16013	29	10	258.41	13.25
Capsules				
NM	30	20	378.78	30.06
NM	31	250	378.87	23.29

According to Figure 3, only three samples are within the limit of variation, with results similar than the labeled concentration of the seized material. Also, inaccuracy of the informed concentration of stanozolol has been found, since some of the samples exhibited double or half the concentration indicated on the package. Moreover, it was noticed a significant heterogeneity of the content in 28 seized samples of stanozolol, and only 3 samples are within the concentration range regulated by the American Pharmacopeia (USP 43) (21).

Figure 3. Control graph of the variation of stanozolol according to the limits established by the 32nd edition of the American Pharmacopeia. The dotted lines indicated the limits in which the concentration should be ranging.

According to the results, 90.3% of the samples appear to be adulterated. Of this percentage, 65% exhibited higher concentrations than those indicated on the package. In this study, the presence of other non-related substances has not been observed. These results exhibit correlation with other studies conducted in Brazil. Berneira et al. observed 25% of counterfeit and 62.5% of adulterated formulations of several AAS including stanozolol (26). Ribeiro et al. observed three main types of adulteration of AAS: absence of the active ingredient, presence of other compounds and concentrations below those indicated. Stanozolol injectable solutions presented contents below the half of the declared on the label. Moreover, the capsules have shown no presence of stanozolol and the existence of other compounds (14). Neves et al. has shown that 53% of the AAS were counterfeit including stanozolol samples that did not present the active ingredient and exhibited lower or higher amount than the indicated (13). The main difference between this study and other data already published consisted of the adulterated samples evaluated exhibited much higher concentrations than those declared on the label. This discrepancy can be correlated with the potential increase of the AAS side effects already described in the literature. In addition, this fact can contribute in intoxication and overdosage processes in athletes, bodybuilders and gym attenders, considering the multiple cycles and drugs used by these public.

Conclusions

This work successfully established a new LC-QTOF-MS method for stanozolol analysis. This method aimed at contributing to the identification and quantitation of the anabolic/androgenic steroid stanozolol frequently seized by federal inspection agencies. Quality control analyses were performed in 31 samples seized by the Secretariat of the Federal Revenue of Brazil by the evaluation of the means and individual weights of tablets and capsules and the content measurement. It was shown that 16.13% of samples reproved the quality control for weight and 90.3% were adulterated. Considering only the adulterated samples, 65% have shown a higher concentration than the one indicated on the label. This study presents concerning data about the quality of AAS acquired in black markets (e.g. academies and/or internet) which can be responsible for several side effects and intoxications.

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Conflict of interest

The authors declare no conflicts of interest.

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